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THE NOVEL PREPARATION OF HYDROTREATING CATALYSTS
USING MECHANOCHEMICAL ACTIVATION OF ALUMINUM NITRATE
AND NIMO COMPLEXES

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The hydrodesulfurization (HDS) process continuously faces strict regulations on sulfur content in liquid fuels on one hand and diminution of available low sulfur feedstock on the other hand. Apart from more severe process condition or major investments into plants, more efficient catalysts represent convenient way how to solve this challenge. The conventional hydrodesulfurization catalysts either γ -Al₂O₃-supported CoMo, NiMo, NiW or unsupported NiMoW sulfides are therefore comprehensively studied in the recent literature. The support is widely modified or replaced by other oxides or non-oxide materials and deposition of the active phase is thoroughly investigated. We have recently reported on novel preparation of γ -Al₂O₃ support by mechanochemical activation of aluminum nitrate hydrate and ammonium carbonate mixture in an agate mill with pestle followed by calcination.

The aim of the present work is to explore optimum preparation route of NiMo catalysts for the novel mechanochemically prepared Al₂O₃ to achieve high HDS activity. Specifically, the novelty of this report lays in comparison of deposition of two type of NiMo complexes: i) the NTA NiMo complex formed by dissolution of (NH₄)₆Mo₇O₂₄, Ni(NO₃)₂ and NTA and ii) the Anderson-type heteropolyoxymolybdate (NH₄)₄Ni(OH)₆Mo₆O₁₈ (HPO) complex. The conventional sequential impregnation using (NH₄)₆Mo₇O₂₄ (first) and Ni(NO₃)₂ (second) is studied as the reference method. The prepared catalysts in their sulfided forms are evaluated by temperature programmed reduction (TPR), nitrogen physisorption, oxygen chemisorption, x-ray photoelectron spectroscopy (XPS), Raman spectroscopy and their activities are determined in reaction of 1-benzothiophene HDS at 360 °C and 1.6 MPa.

The complexes of NiMo with NTA or (NH₄)₄Ni(OH)₆Mo₆O₁₈ were found to be feasible precursors for preparation of catalysts highly active in hydrodesulfurization reaction of 1-benzothiophene. The NTA assisted deposition resulted in the catalyst with low metal-support interaction and 1.5 higher activity than the conventionally prepared NiMo counterpart. The most promising method of NiMo catalyst preparation for the studied γ -Al₂O₃ was the deposition of (NH₄)₄Ni(OH)₆Mo₆O₁₈. An optimal interaction between this complex and the new alumina, manifested by the presence of MoO₄²⁻ or MoO_xS_y species, was found to be beneficial and determining factor of about 1.9 and 1.4 higher HDS activity in comparison to the conventionally prepared NiMo and reference commercial NiMo counterparts, respectively.

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