



národní  
úložiště  
šedé  
literatury

## **Photocatalytic and Photolytic Degradation of 4-Nonylphenol in Two Types of Batch Reactors**

Spáčilová, Lucie  
2014

Dostupný z <http://www.nusl.cz/ntk/nusl-174817>

Dílo je chráněno podle autorského zákona č. 121/2000 Sb.

Tento dokument byl stažen z Národního úložiště šedé literatury (NUŠL).

Datum stažení: 10.04.2024

Další dokumenty můžete najít prostřednictvím vyhledávacího rozhraní [nusl.cz](http://nusl.cz).

# Photocatalytic and Photolytic Degradation of 4-Nonylphenol in Two Types of Batch Reactors

*Student: Ing. Lucie Spáčilová*  
*Supervisor: Ing. Olga Šolcová, DSc.*

4-nonylphenol belongs to the group of the environmental estrogens that act as the hormonal system of the body. It is present in waste water and sewage treatment plants fail to decrease the content of this compound.

The degradation of 4-nonylphenol was measured in two types of batch reactors: Beaker with  $\text{TiO}_2$  bed (Fig. 1) and Erlenmeyer flask with  $\text{TiO}_2$  on glass beads in suspension (Fig. 2). The initial concentration of 4-nonylphenol was ranged between 8 to 12 ppm.

$\text{TiO}_2$  thin layers were prepared by the sol-gel process controlled in a reverse micellar environment.  $\text{TiO}_2$  was synthesized by the addition of Titanium (IV) isopropoxide into formed inverse micellar solution made of cyclohexane, nonionic surfactant Triton X-114 and distilled water. The molar ratio of cyclohexane/Triton X-114/water/ $\text{Ti}(\text{OC}_3\text{H}_7)_4$  was kept at 11/1/1/1 (volume ratio TX-114/cyclohexane = 0.49). Prepared solution was stirred intensively for homogenization and formation of inverse micelles. Then liquid Titanium (IV) isopropoxide was regularly dropped to the micellar solution during the vigorous stirring.  $\text{TiO}_2$  films were deposited by the four cycles of the dip-coating method. Among the single dip-coating cycles the samples were thermally treated by calcination at 400 °C for 4 hours with the temperature

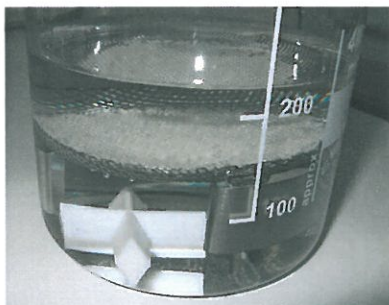


Figure 1. Batch reactor – Beaker with  $\text{TiO}_2$  bed



Figure 2. Batch reactor – Erlenmeyer flask

ramp 1 °C/min in the air flow in a muffle furnace. TiO<sub>2</sub> photocatalyst in the anatase crystalline form possesses surface area 70 m<sup>2</sup>/g with the small content of micropores ( $V_{micro} = 15 \text{ mm}^3/\text{g}$ ) and the layer thickness 340 nm. For photocatalyst activation the medium pressure mercury lamp was used.

Experimental data show involvement of two degradation processes; photocatalysis together with photolysis. Photolytic process generates intermediates with high toxicity and estrogenic activity and therefore is desirable to minimize it. For suppressing of the photolysis process the special glass filter was applied to eliminate UV C and 80 % of UV B radiation. The endocrine disruptor degradation activity was compared together with selectivity for both arrangements with respect to the possible applications.